# New Compounds: 4-Substituted 5,6-Dihydro-2-o-hydroxyphenyl-4H-1,3,4-oxadiazine-5-ones, Potential Psychopharmacological Drugs

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Abstract  $\Box$  The synthesis of 16 derivatives of a 1,3,4-oxadiazine, each of which may be envisioned as forming a tricyclic structure, was carried out by 4-alkylation and further reaction with a secondary amine. The compounds are proposed as potential psychopharmacological drugs.

**Keyphrases** □ 5,6-Dihydro-2-o-hydroxyphenyl-4H-1,3,4-oxadiazine-5-ones, 4-substituted—synthesized as potential psychopharmacological drugs □ Psychopharmacological agents, potential synthesis of 4-substituted 5,6-dihydro-2-o-hydroxyphenyl-4H-1,3;4-oxadiazine-5-ones □ Oxadiazines—synthesis of 4-substituted 5,6-dihydro-2-o-hydroxyphenyl-4H-1,3,4-oxadiazine-5-ones

As a part of a program of synthesis of potential new psychopharmacological drugs derived from little explored heterocycles<sup>1</sup>, several derivatives were obtained from a previously described compound, 5,6dihydro-2-o-hydroxyphenyl-4*H*-1,3,4-oxadiazine-5one (4). An outstanding feature of these oxadiazines is their potential ability to form a tricyclic plane structure due to the ease with which a stable hydrogen bond may form between the o-phenolic group of the aromatic substitution on carbon 2 and the position 3 nitrogen, in effect producing a cryptophenol.

The N-4-alkylations involved two different approaches: direct alkylation, and a two-step process consisting of direct alkylation to yield an  $\alpha$ - or  $\beta$ -chloroalkyl derivative followed by treatment of the 4-chloroalkyloxadiazine with a secondary amine.

#### EXPERIMENTAL<sup>2</sup>

5,6-Dihydro-2-o-hydroxyphenyl-4-(2-chloroethyl)-4H-1,3,4oxadiazine-5-one (III)—A suspension of 0.01 mole (1.92 g) of 5,6-dihydro-2-o-hydroxyphenyl-4H-1,3,4-oxadiazine-5-one in 35 ml of acetone was dissolved by adding, with stirring, 3.5 ml of 10% sodium hydroxide. After a few minutes the sodium salt crystallized; 0.026 mole (2.2 ml) of 1-bromo-2-chloroethane was then added and the mixture was refluxed for 2 hr. On standing overnight, the product crystallized (Table I).

5,6-Dihydro-2-o-hydroxyphenyl-4-(3-chloropropyl)-4H-1,3,4-oxadiazine-5-one (IV)—This compound was obtained by using the procedure for III with 1-bromo-3-chloropropane (Table I).

4-Alkyl Oxadiazines: General Procedure—Method A—To a suspension of 0.55 g of sodium methoxide in 35 ml of acetone was added, with stirring, 0.01 mole of 5,6-dihydro-2-o-hydroxyphenyl-



Table I—Physical	Constants of	Compounds I-XVI
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Com- pound	R	Method	Melting Point	Yield, %	Crystallization Solvent	Analysis, %	
						Calc.	Found
Ţ	-CH <sub>2</sub> CH <sub>3</sub>	A	78°	75	Methanol	C 60.00 H 5.40 N 12.73	59.92 5.38 12.70
II	$-CH_2CH_2CH_3$	Α	<b>9</b> 3°	67	Ethanol-water	C 61.53 H 5.98 N 11.96	61.50 5.93 11.95
111	CH <sub>2</sub> CH <sub>2</sub> Cl		164–165°	63	Dioxane	C 51.86 H 4.32 N 11.00	$52.00 \\ 4.40 \\ 10.85$
IV	$-CH_2CH_2CH_2Cl$		168169°	74.6	Dioxane	C 53.63 H 4.84 N 10.42	53.50 4.78 10.39
V	—СН2-	Α	115°	62	Ethanolwater	C 68.08 H 4.96 N 9.92	68.00 4.99 9.89
VI	-CH_CH_O	Α	110°	50	Cyclohexane–ethanol	C 65.38 H 5.13 N 8.97	65.35 5.15 8.96
VII	-CH <sup>2</sup> CH <sup>3</sup> O-Cl	A	133-139°	59	Ethanol	C 58.87 H 4.32 N 8.08	58.85 4.40 7.98
VIII	-CH <sub>2</sub> CH <sub>2</sub> O-CH <sub>3</sub>	Α	102–10 <b>9</b> °	60	Methanol	C 66.25 H 5.52 N 8.58	$\begin{array}{r} 66.21\ 5.49\ 8.55 \end{array}$

<sup>1</sup>Little information is available about 1.3,4-oxadiazines with pharmacological activity (1-3), and nothing is available concerning the corresponding 5-ones. <sup>2</sup> Melting points (Buchi apparatus) are uncorrected; IR spectra recorded on Ferkin-Elmer 137B Infracord; spectral data consistent with structure assigned; OH band, 3000 cm<sup>-1</sup>; C=O, 1670 cm<sup>-1</sup> (potassium bromide).

Com-			Melting			Analysis, %	
pound	R	Method	Point	Yield, %	<b>Crystallization Solvent</b>	Calc.	Found
IX	-CH <sub>2</sub> CH <sub>2</sub> -N(CH <sub>3</sub> ) <sub>2</sub> . HCl	В	196°	54	Methanol-benzene	C 52.08 H 6.01 N 14.02	51.98 6.12 13.97
Х	$\begin{array}{c} -\mathbf{C}\mathbf{H}_{2}\mathbf{C}\mathbf{H}_{2} - \mathbf{N}(\mathbf{C}_{2}\mathbf{H}_{5})_{2} \cdot \\ \mathbf{H}\mathbf{C}\mathbf{l} \end{array}$	В	199°	58	Methanol	C 54.96 H 6.71 N 12.82	$55.05 \\ 6.75 \\ 13.00$
XI	-CH <sub>2</sub> CH <sub>2</sub> -N · HCl	В	205–206°	35	Methanol-benzene	C 55.29 H 6.14 N 12.90	$55.18 \\ 6.16 \\ 12.82$
XII	CH <sub>2</sub> CH <sub>2</sub> N · HCl	В	231–232°	79.4	$\mathbf{E}$ thanol	C 56.55 H 6.48 N 12.37	56.47 6.55 12.26
XIII	-CH <sub>2</sub> CH <sub>2</sub> -NO · HCl	В	256–258°	60	Methanol-benzene	C 52.70 H 5.85 N 12.30	52.72 6.01 12.42
XIV		В	177~178°	50.3	Methanol	C 56.55 H 6.48 N 12.37	56.65 6.50 12.28
xv	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> -N · HCl	В	204–205°	35	Methanol	C 59.38 H 6.98 N 9.31	<b>59</b> .29 7.00 9.30
XVI	-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> -NO·HCl	В	223–224°	70.6	Methanol	C 54.01 H 6.19 N 11.81	53.96 6.30 11.90

4H-1,3,4-oxadiazine-5-one. After a few minutes the sodium salt crystallized. Then 0.02 mole of the appropriate bromo derivative was added, and the mixture was refluxed for 5 hr and filtered. The solvent was evaporated *in vacuo* and the solid residue was crystallized from an appropriate solvent (Table I).

Method B—A mixture of 0.01 mole of III or IV and 0.06 mole of the corresponding secondary amine was heated at 80° in an oil bath for 6 hr and filtered, and the excess base was evaporated in vacuo at 80°. The residue was dissolved in benzene and the hydrochloride was precipitated by adding a solution of anhydrous hydrochloric acid in ether. It was recrystallized from an appropriate solvent (Table I).

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